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CHEMISTRY: A Study of Matter © 2004, GPB 4.18 III. An element has the electron distribution $1s^22s^22p^63s^23p^64s^23d^{10}4p^4$. Use this information to answer the questions.
Electron Notations Review - ScienceGeek.net
www.sciencegeek.net/Chemistry/taters/Unit2ElectronNotations.htm Which of the

following is the correct electron configuration notation for the element

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Practice: Write electron distributions and do the orbital notation for the following:
1. P : 2. Ca: Only do the electron distributions for the following: 1. Co: 2. Eu: 3. Tc:
orbitals Arows quantum elctron n l m s n enrgy l orbital ua nt mbers energy
sublevel orbital

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Using periodic table to find electron distributions: Noble Gas Distributions – Find noble gas on end of row before the element and put its symbol in brackets. Use diagonal rule to write electron distribution of Sn (atomic no. 50): ____ Use noble gas distribution to write electron distribution of Sn: ____ The Chemistry Quiz CR1. CR2. 1. 2

3-19a,19b-Electron Distributions Wkst-Key

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