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cen84959_ch08.qxd 4/20/05 4:05 PM Page 461. Chapter 8 Assumptions 1 This is a steady-flow process since there is no change with time at any point and thus $m\dot{C}V = 0$, $E\dot{C}V = 0$, and $X\dot{C}V = 0$. 2 The kinetic

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Avogadro's law states that $V \propto n$, which can be rearranged to give $\frac{V_1}{n_1} = \frac{V_2}{n_2}$. Since a is a constant, an alternative representation is $V_1 V_2 = a n_1 n_2$ where V_1 is the volume of n_1 moles of O_2 gas and V_2 is the volume of n_2 moles of O_3 gas. In this case we have Ar CH₄ n_1 0.50 mol n_2 0.33 mol V_1 12.2 L V_2 ?

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Applicants must possess N2/N3 technical qualification with relevant subjects, i.e.:
Auto Electrician (Science and Maths, Electro-technology/ Electrical Trade Theory or
Industrial electronics) Diesel Mechanic (Diesel Trade Theory, Mathematics,
Engineering Drawing, Science)

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